# Chemical and toxicological characterization of chemicals released from plastic polymers under ultraviolet light

**Matthew MacLeod**<sup>1</sup>, Berit Gewert<sup>1</sup>, Merle Plassmann<sup>1</sup>, Oskar Sandblom<sup>1</sup>, Christoph D. Rummel<sup>2</sup>, Hans Peter H. Arp<sup>3,4</sup>, Beate I. Escher<sup>2,5</sup> & Annika Jahnke<sup>2</sup>

1 – Department of Environmental Science, Stockholm University, Sweden

- 2 Department of Bioanalytical Ecotoxicology and Department of Cell Toxicology, Helmholtz Centre for Environmental Research-UFZ, Leipzig, Germany
- 3 Department of Environmental Engineering, Norwegian Geotechnical Intsitute (NGI), Oslo, Norway
- 4 Department of Chemistry, Norwegian University of Science and Technology (NTNU), Trondheim, Norway
- 5 Center for Applied Geoscience, Eberhard Karls University Tuebingen, Tuebingen, Germany

Chemical and toxicological characterization of chemicals released from plastic polymers under ultraviolet light

Hypothesis

Plastic polymers themselves are the source of chemical contaminants that could pose a risk to the environment





Chemical and toxicological characterization of chemicals released from plastic polymers under ultraviolet light Goals

- 1. Develop a lab procedure to artificially weather plastics under UV-light
- 2. Identify chain-scission degradation products of different polymers
- 3. Screen leachates from weathering plastics for toxicity & compare to pristine plastic



JPI OCEANS

### 1. Artificial weathering of plastic



- The "weathering wheel"
- Air-cooled UV lamp in center
- 5 g of plastic material & 250 mL water
- 5 days  $\rightarrow$  ~ 510 days of sunlight
- Blank samples & dark controls



- Solid phase extraction (SPE) of water
- Analysis via LC-HRMS
  - LC: C18 column run with a gradient of H<sub>2</sub>O and ACN
  - HRMS: Q Exactive HF Orbitrap
  - Full scan mode + data dependent MS2 fragmentation
- Data processing: Compound Discoverer 2.1
  - Peak picking and alignment between samples
  - Molecular formula prediction



Signals detected in leachates of pre-production polyethylene (Goodfellow, Gmbh)

965



1 Molecular formula: C<sub>x</sub>H<sub>y</sub>H<sub>z</sub> ≤ 5ppm

Signals detected in leachates of pre-production polyethylene (Goodfellow, Gmbh)

965 256























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Letter

#### Identification of Chain Scission Products Released to Water by Plastic Exposed to Ultraviolet Light

Berit Gewert, Merle Plassmann, Oskar Sandblom, and Matthew MacLeod\*®

Department of Environmental Science and Analytical Chemistry, Stockholm University, Svante Arrhenius väg 8, 11418 Stockholm, Sweden

S <u>Supporting Information</u>

**ABSTRACT:** Buoyant plastic in the marine environment is exposed to sunlight, oxidants, and physical stress, which may lead to degradation of the plastic polymer and the release of compounds that are potentially hazardous. We report the development of a laboratory protocol that simulates the exposure of plastic floating in the marine environment to ultraviolet light (UV) and nontarget analysis to identify degradation products of plastic polymers in water. Plastic pellets [polyethylene, polypropylene, polystyrene, and poly-(ethylene terephthalate)] suspended in water were exposed to a UV light source for 5 days. Organic chemicals in the water were concentrated by solid phase extraction and then analyzed



by ultra-high-performance liquid chromatography coupled to high-resolution mass spectrometry using a nontarget approach with a C18 LC column coupled to a Q Exactive Orbitrap HF mass spectrometer. We designed a data analysis scheme to identify chemicals that are likely chain scission products from degradation of the plastic polymers. For all four polymers, we found homologous series of low-molecular weight polymer fragments with oxidized end groups. In total, we tentatively identified 22 degradation products, which are mainly dicarboxylic acids.  22 chain-scission degradation products of polyethylene, polystyrene and polyethylene terephthalate identified.

Mostly dicarboxylic acids.

 515 additional possible degradation products of these polymers identified by exact mass and molecular formula CxHyOz.



 Exploratory experiments with artificial weathering of diverse plastic materials

• Toxicity of leachates to *nitrocra spinipes* could either increase of decrease as a result of weathering

Bejgarn, MacLeod, Bogdal, Breitholtz. Toxicity of leachates from weathering plastics: An exploratory study. *Chemosphere*. 132 114-119 **2015**.

Cell-based bioassays

Department of bioanalytical ecotoxicology, Helmholtz Centre for Environmental Research – UFZ, Leipzig



recombinant cells

**Assay** AhR AREc32 ΡΡΑRγ

High-throughput bioassays

Endpoint Xenobiotic metabolism Antioxidant response element Peroxisome proliferating activity













 Positive AhR response (xenobiotic metabolism) for the Ewaste and keyboard controls

• No response for the pre-production plastics





- Antioxidant response from all leachates
  - Stronger for UVweathered material
- Strongest (non-positive control response) for
  weathered polyethylene and polypropylene.





Peroxisome
proliferating response
for all leachates

 Strong response for UV-weathered polyethylene.



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#### References

#### Pathways for degradation & identification of chain scission products

Gewert, B.; Plassmann, M.; MacLeod, M., Pathways for degradation of plastic polymers floating in the marine environment. *Environmental Science: Processes & Impacts* **2015** 17 (9) 1513-1521.

Gewert, B.; Plassmann, M.; Sandblom, O.; MacLeod, M. Identification of chain scission products released to water by plastic exposed to ultraviolet light. *Environmental Science & Technology Letters* **2018**, *5* (5), 272–276.

#### Toxicity in-vivo and in cell-based bioassays

Bejgarn, S.; MacLeod, M.; Bogdal, C.; Breitholtz, M., Toxicity of leachate from weathering plastics: An exploratory screening study with *Nitocra spinipes*. *Chemosphere* **2015** 132 114-119.

Rummel, C. D.; Escher, B. I.; Sandblom, O.; Plassmann, M. M.; Arp, H. P. H.; MacLeod, M.; Jahnke, A. Effects of leachates from UV-weathered microplastic in cell-based bioassays. *Environmental Science & Technology* **2019**, *53* (15), 9214–9223.







- "Iceberg modeling" of the PPARγ response
- Response of pure linear mono- and di-carboxylic
  acids explains up to 42% of the response for UVweathered polyethylene



Pure samples of reference plastic materials have distinct fingerprints



But in experiments with mixtures of pure plastics we cannot find the same "fingerprint"!









